

The graphic features a blue wavy background. In the top left is the American Water Works Association logo and name. The central text reads 'CONTAMINANTS OF CONCERN: Managing Lead and Manganese' and 'WEBINAR SERIES'. Below this is the date and time: 'May 26th, 2020 | 1:00 – 2:30p.m (Mountain) Strategies for Understanding and Managing Risk from Lead'. The background includes a molecular model and a portion of the periodic table.

American Water Works Association
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CONTAMINANTS OF CONCERN:
Managing Lead and Manganese

WEBINAR SERIES

May 26th, 2020 | 1:00 – 2:30p.m (Mountain)
Strategies for Understanding and Managing Risk from Lead

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Webinar Moderator



Michelle DeHaan
Water Quality & Treatment Manager
Park City Municipal Corporation

Michelle De Haan has 28-years of experience as a drinking water quality and treatment specialist, over the last 9-years as Park City Utah's Water Quality & Treatment Manager. She was an AWWA Planning Committee Member for the 2020 International Symposium on Inorganics, is a Past AWWA Trustee of Water Science & Research Division, is a past AWWA Chair of Inorganic Contaminant Committee, and was the 2018 AWWA Water Science and Research Division Emerald Erlenmeyer Awardee.

3



3

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5

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6

Panel of Experts



Sandra Kutzing
Principal Engineer
CDM Smith



Carol Rego
Water Supply and
Treatment Specialist
CDM Smith



Sophie Manley
Sanitary Engineer III
City of Chicago
Department of
Water Management



Kiran Udayakumar
Water Quality Engineer
Arcadis



John Tobiason
Professor and
Department Head of
Civil and
Environmental
Engineering
University of
Massachusetts at
Amherst

7



7

Agenda

- I. Factors that Influence the Effectiveness of POU Filters to Remove Lead**
Sandra Kutzing, CDM Smith, Carol Rego, CDM Smith
- II. Managing Millions of Data Points – Administering Chicago’s Free Lead Testing Program**
Sophie Manley, City of Chicago Water, Kiran Udayakumar, Arcadis
- III. Assessment of Lead in Childcare and School Drinking Water**
John Tobiason, University of Massachusetts Amherst

Time Permitting – Q&A

Enter your **question** into the **question pane** at the lower right-hand side of the screen.

Please specify to whom you are addressing the question.

8



8

Factors that Influence the Effectiveness of POU Filters to Remove Lead

Carol A. Rego, P.E.
Sandra L. Kutzing, P.E.

May 26, 2020



**CDM
Smith**

9



Overview

- Newark's Corrosion Control Background
- POU Filter Study
 - Why Test the Filters?
 - Full Program Roll-out
 - Study Results
 - Conclusions and Recommendations

10

10



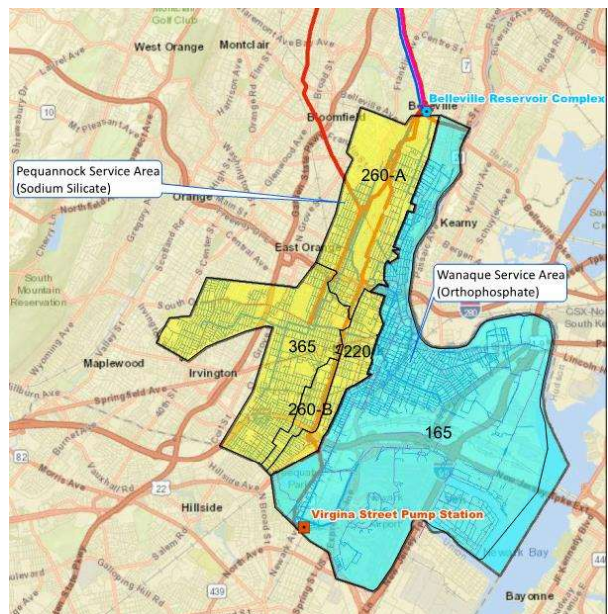
Newark's Corrosion Control Background



11

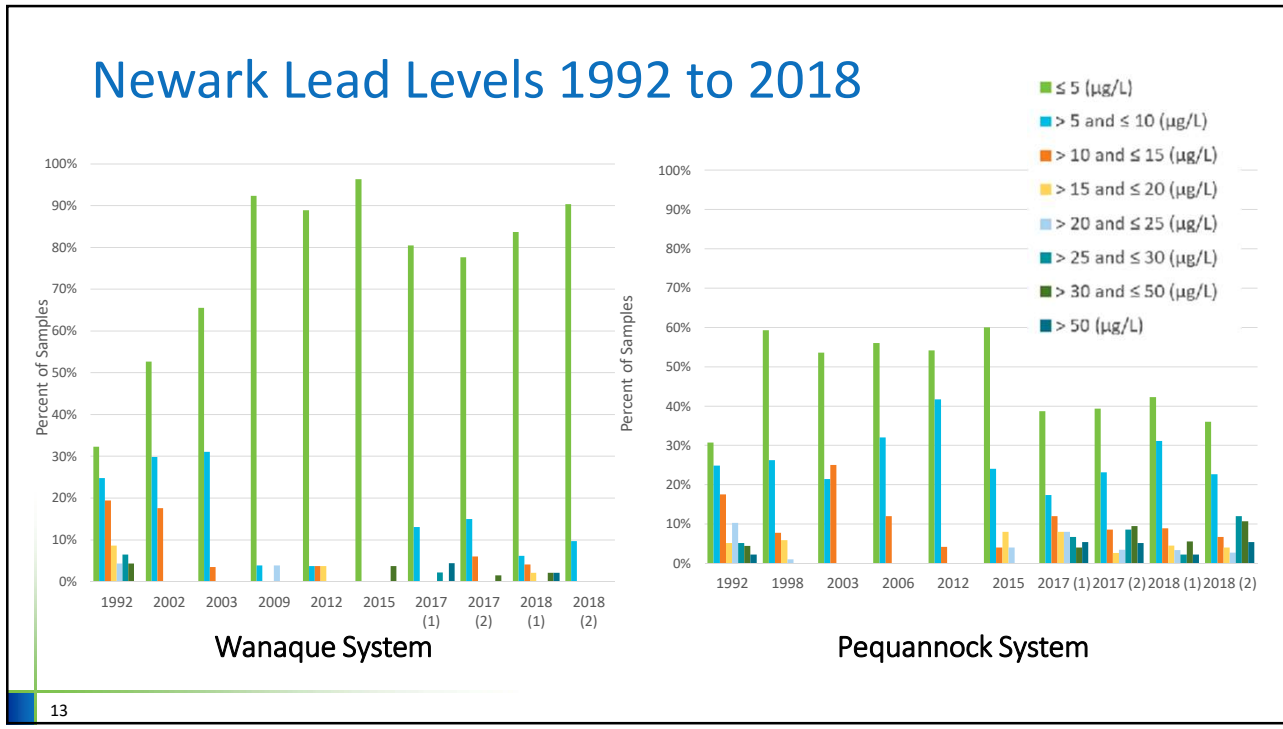
City of Newark, NJ

- 300,000 customers
- 60 million gallons per day
- Approximately 20,000 LSLs
- Two supplies with different corrosion control treatment (CCT)
 - Pequannock: sodium silicate (now zinc orthophosphate)
 - Wanaque: zinc orthophosphate



12

12



13

Typical Conditions for <15 ppb Lead Action Level in Northeast Waters

Orthophosphate Inhibitor

Carbonate (pH/Alk Adjustment)

pH 7.2-7.8

Ortho PO₄ 0.5-2.0

pH 7.2-7.8

Zinc Ortho PO₄ 0.5-2.0

pH 7.2-7.8

Ortho/Poly PO₄ 1.0-2.0

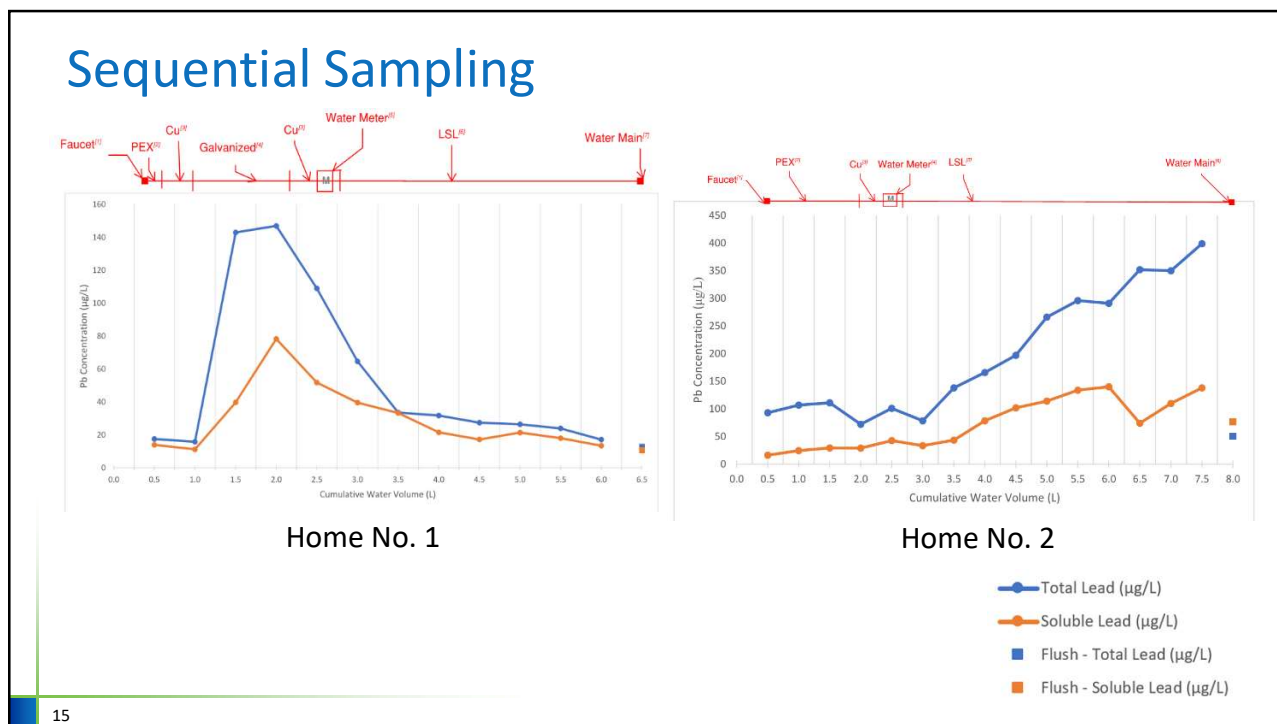
pH 9.4-9.8

Alk 30-40

DIC 4-6

- Newark meeting Lead Action Level with pH 7.0-7.2, sodium silicate
 - Highly suspicious of potential for tetravalent lead (plattnerite, PbO₂) scale
 - But... chlorine residual not excessively high

14



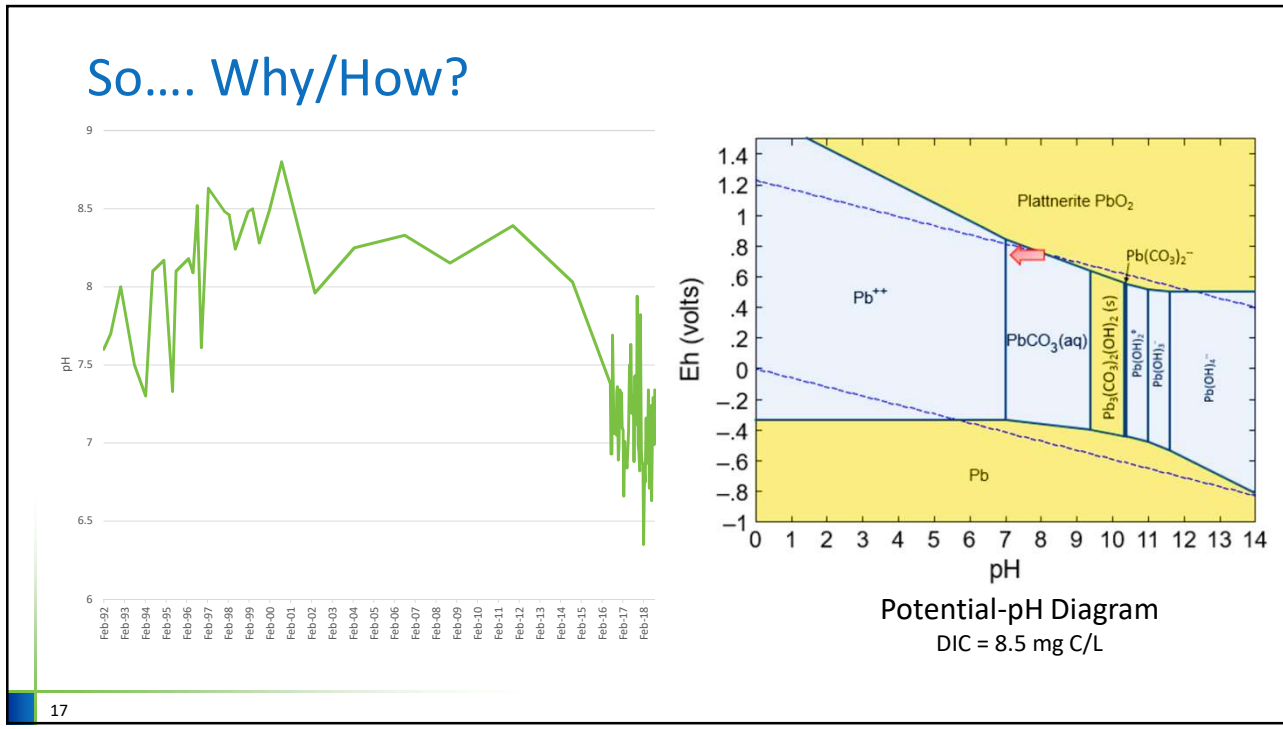
15

Scale Minerology

- Pipes did not contain any crystalline Si-Pb compounds
- Plattnerite (PbO_2) predominant in scale minerology
- Pb(II) carbonates growing up through plattnerite
- Plattnerite not being maintained:
 - Uppermost layer not very robust
 - Porous and easily detachable, therefore not providing a very resilient barrier to lead release

Courtesy EPA Office of Research & Development

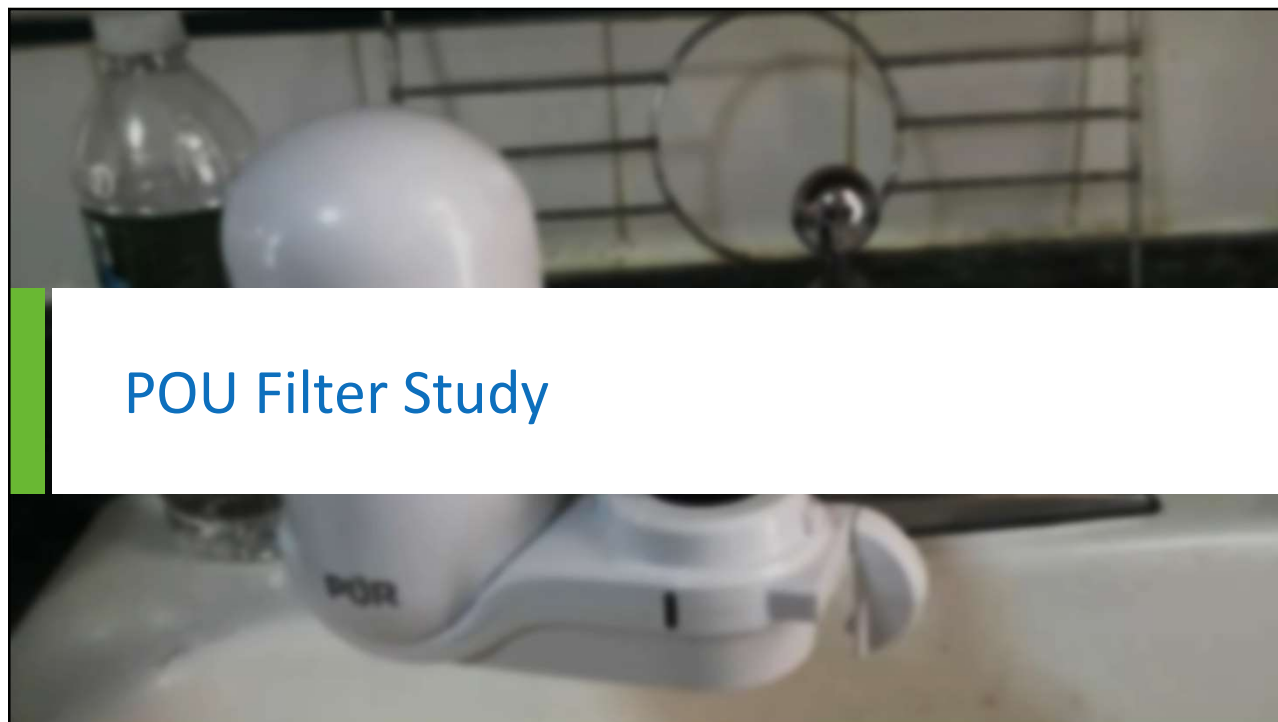
16



17

- ### Public Health Aspects
- pH decreased (above 8 to 7.1 on average) for DBP simultaneous compliance (but silica maintained)
 - Plattnerite protective scale not maintained at lower pH
 - Multiple factors provided evidence of an urgent concern:
 - Theoretical chemistry
 - Sequential sampling
 - Scale mineralogy
 - **Decision** → *immediate* POU filter distribution to Pequannock service area customers with LSLs and lead solder

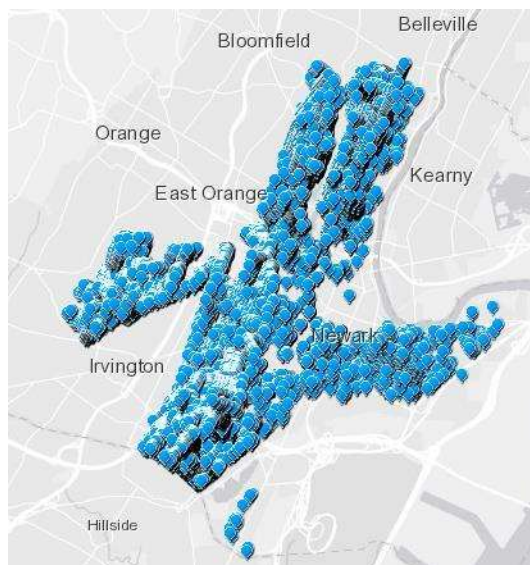
18



19

Newark's Point-of-Use (POU) Filter Distribution

- Over 40,000 filters
- Over 35,000 packages of filter cartridges
- Distributed to homes:
 - In Pequannock with LSLs or lead solder
 - Anywhere in the City that test over 15 ppb



20

20

The Customer Expectation...



21

The Certification...



Reduce lead levels to
10 ppb or less*

**Certification changed to
5 ppb in December 2019*

21

POU Filter Certifications

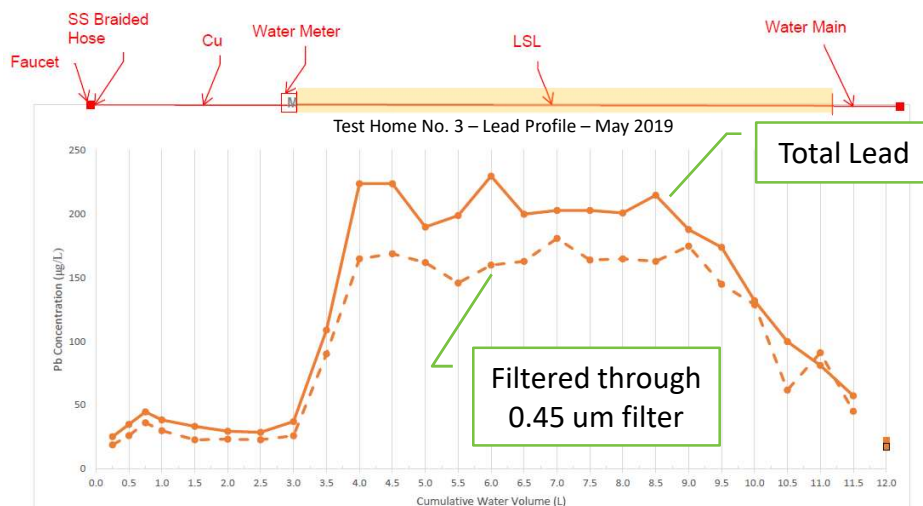
- NSF/ANSI Standard 42 for Class 1 particulates (0.5 to < 1 μm)
- NSF/ANSI Standard 53 for reduction of soluble and particulate lead
- Specific test water is used—2 specific conditions
- 150 ppb of lead in the influent challenge water
- Reduce lead to 10 ppb (now 5 ppb as of December 2019)



22

22

Why Test the Filters?



23

23

Initial Filter Testing Parameters

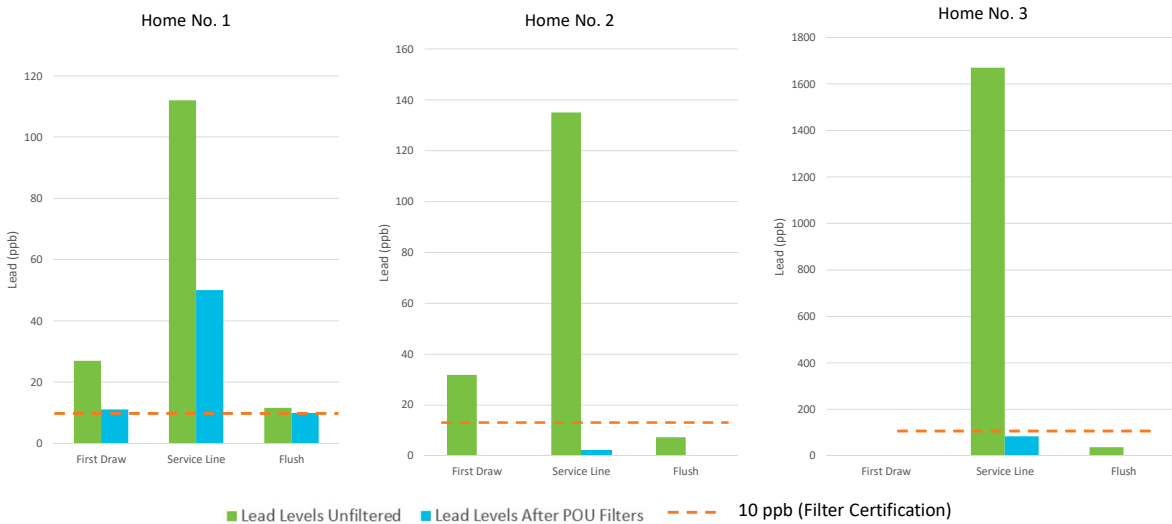
- Three (3) initial POU filters tested in July and August 2019 under extreme conditions
 - 6+ hours of stagnation
 - Targeted water sitting in lead service line
 - Also tested first draw and after a 10 minute flush
 - 500 mL samples collected—both unfiltered and filtered



24

24

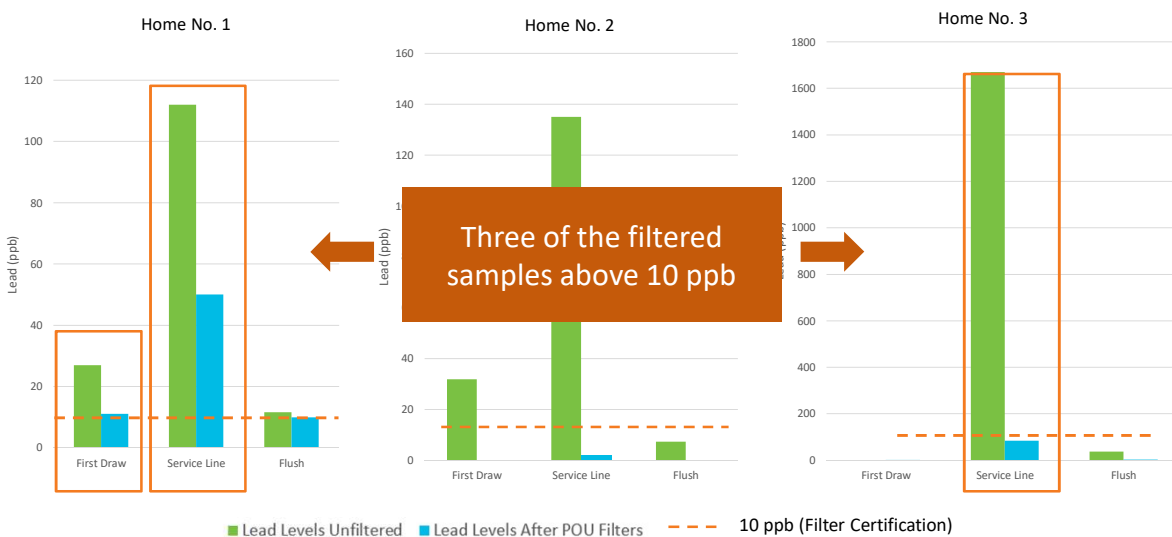
The Results of the Initial 3 Filters Tested



25

25

The Results of the Initial 3 Filters Tested



26

26

Filter Study Roll Out

- Protocol developed with Newark, EPA and DEP
 - Random sampling of first draw, service line, flushed
 - Filters properly installed and maintained
- 20-25 samplers each day for 4 weeks
- 3 labs–Newark, DOH, EPA
- Custom survey app to collect data
- Daily coordination meetings
- Transparency–data posted to SharePoint



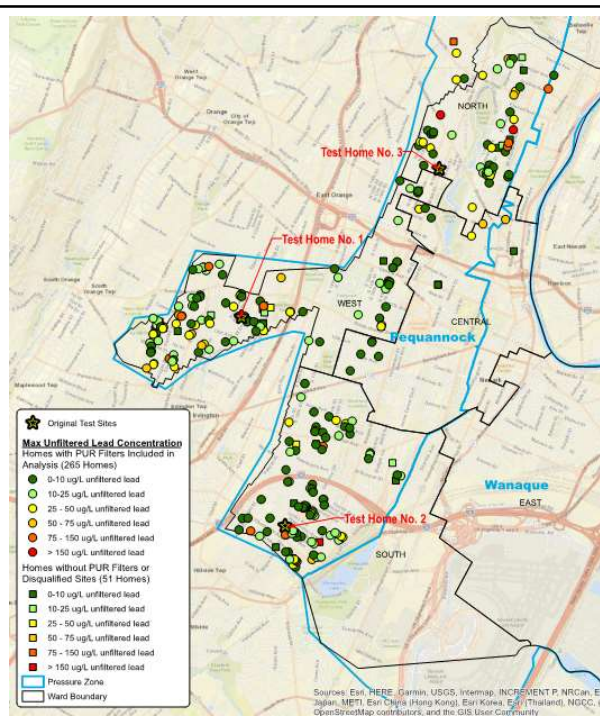
27

27

Filter Study Overview

- 1,573 samples collected
- 337 sampling events
- 316 homes
- 265 PUR filters tested (type distributed by City)

All collected in 4 weeks



28

28

Evaluation of Impacts on Filters

- Proper installation and use of filters
- Unfiltered lead levels – first draw, service line, flushed
- Service line material
- Flushing
- Filter type–faucet, pitcher
- Premise plumbing material
- Recent plumbing changes or construction
- Time since water last used in house and at faucet
- Last time cartridge was replaced
- Filter manufacturer and model
- Cartridge model
- Common filter uses
- Cold and/or hot water usage through filter
- Filter flow rate
- Length of service line

29

29

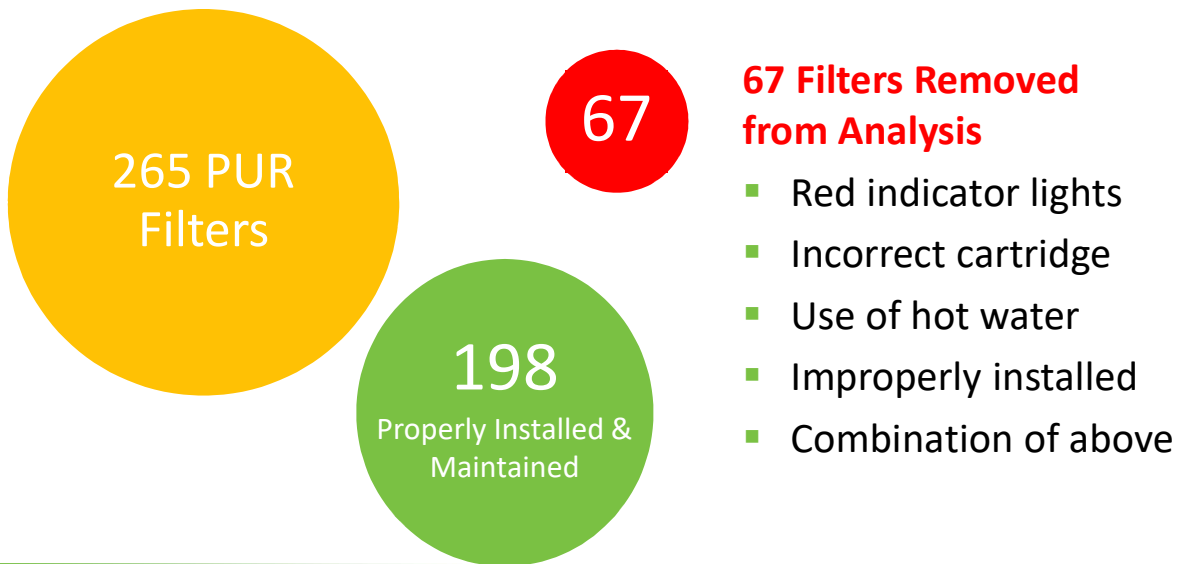
Evaluation of Impacts on Filters

- **Proper installation and use of filters**
- **Unfiltered lead levels – first draw, service line, flushed**
- **Service line material**
- **Flushing**
- **Filter type – faucet, pitcher**
- Premise plumbing material
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- Filter manufacturer and model
- Cartridge model
- Common filter uses
- Cold and/or hot water usage through filter
- Filter flow rate
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30

30

Proper Installation and Use of Filters



31

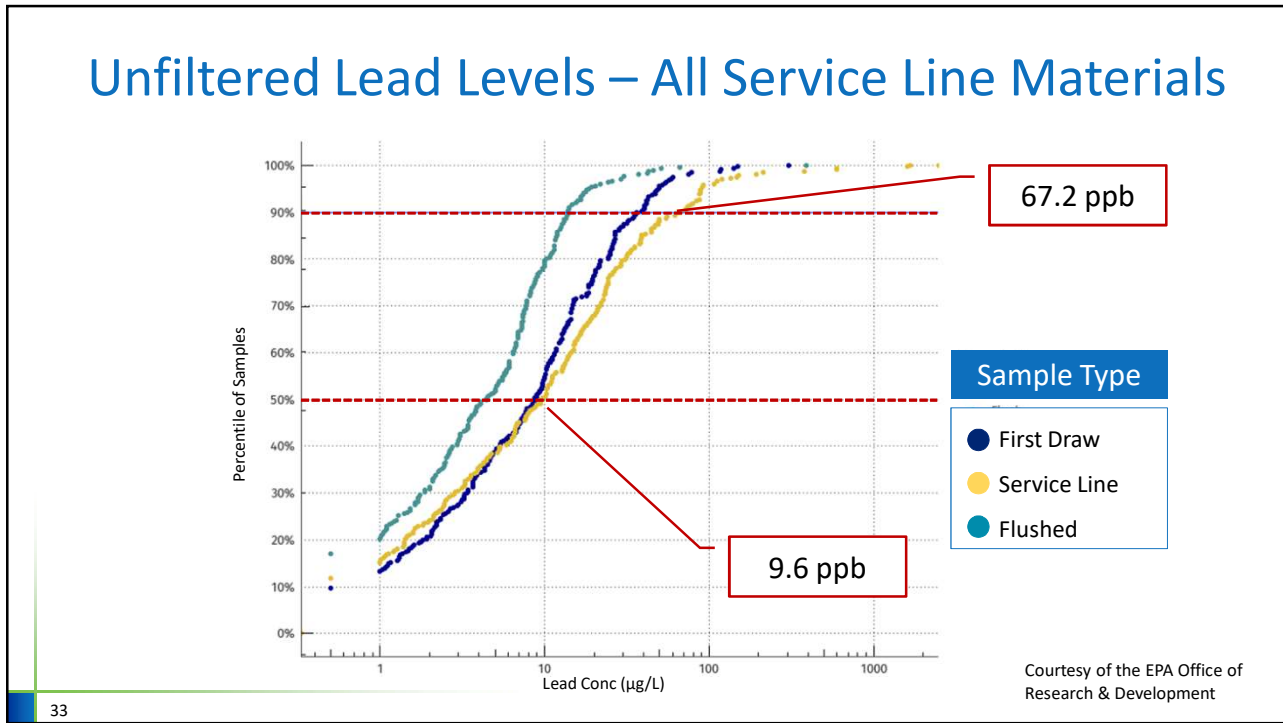
31

Influence of Proper Installation and Use on Filters

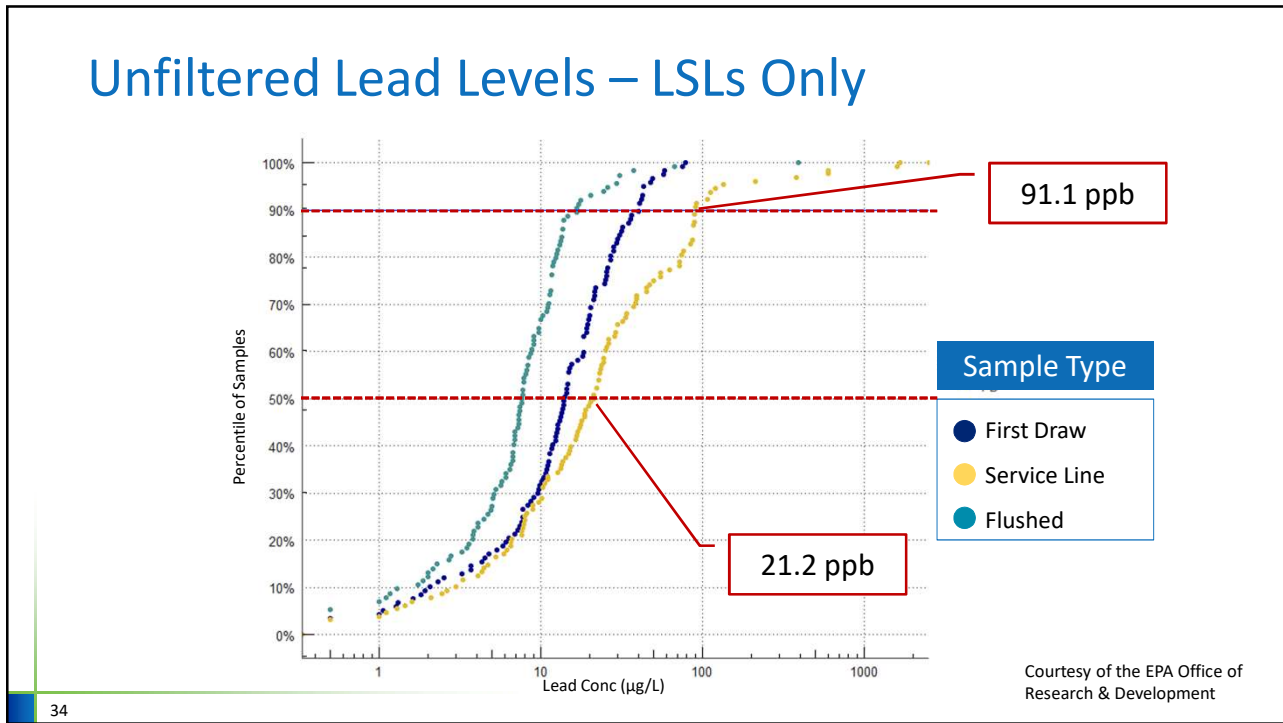
Scenario	No. Filters	% All Filtered Samples Below 10 ppb	% All Filtered Samples Below 5 ppb
All Filters (Provided by Newark)	265	96.6%	90.6%
Filters Properly Installed & Maintained	198	97.5%	91.5%

32

32

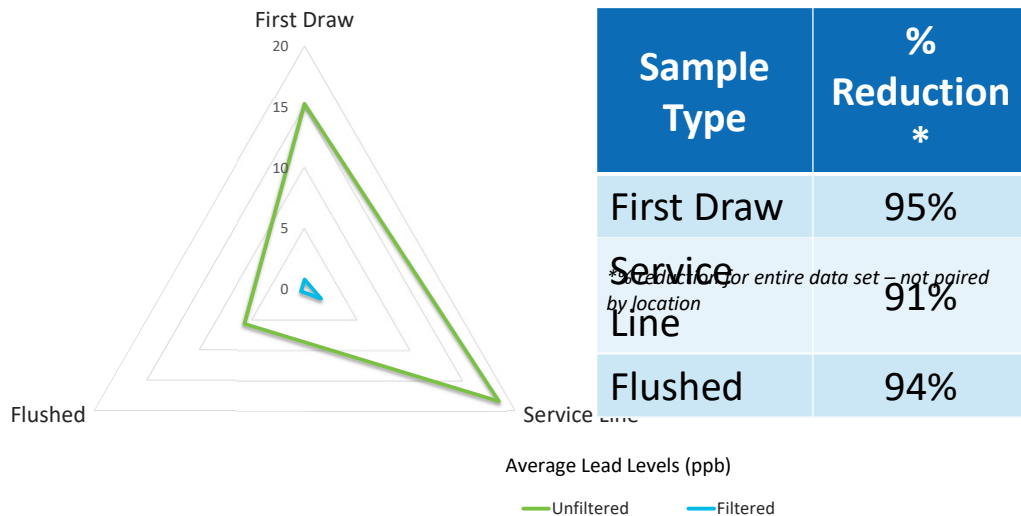


33



34

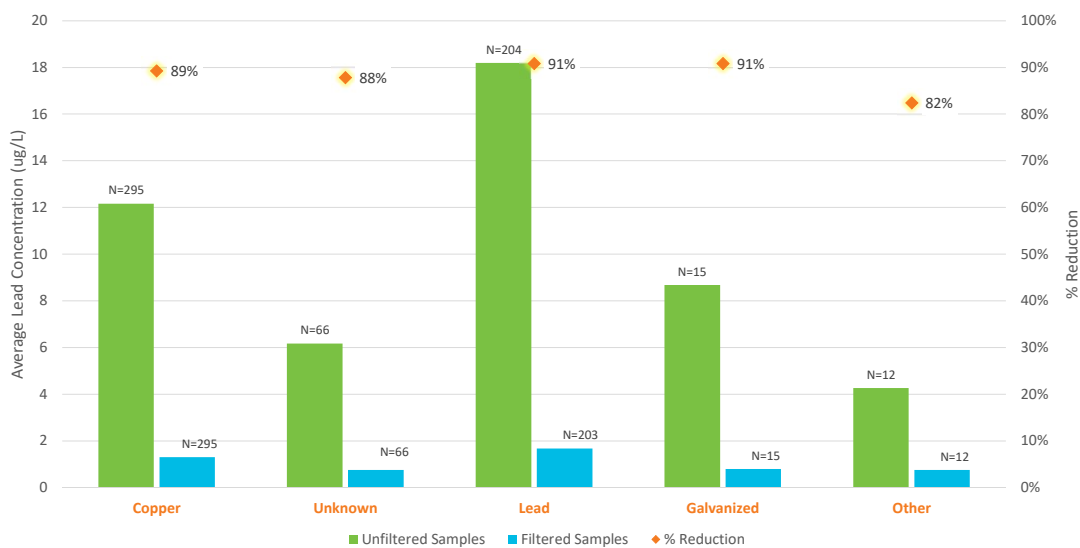
Influence of Unfiltered Lead Levels on Filters



35

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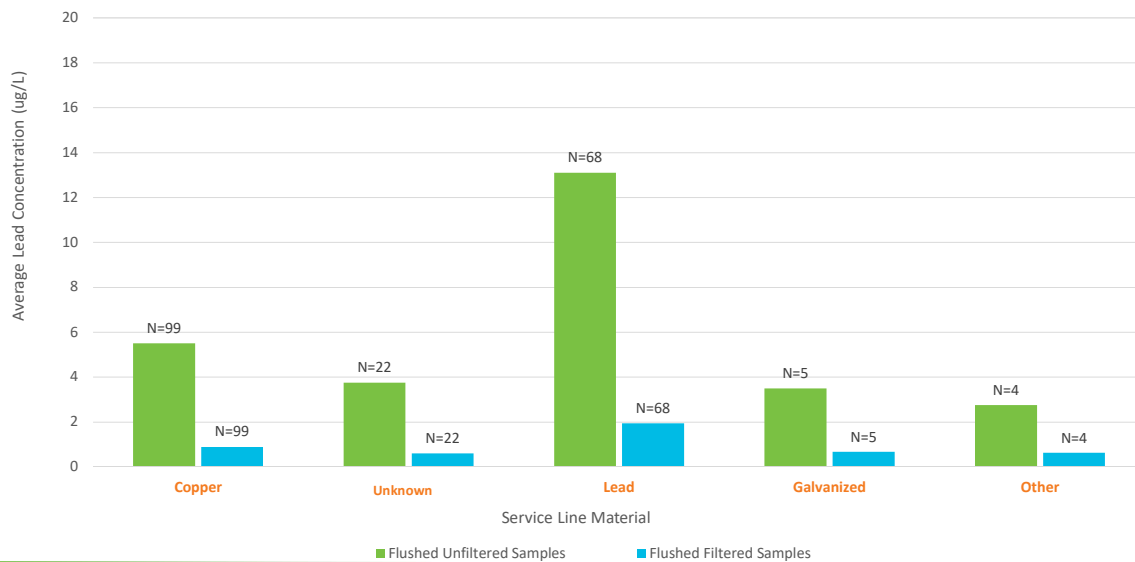
Influence on Service Line Material on Filters



36

36

Influence of Flushing on Filters (5 min flush)



37

37

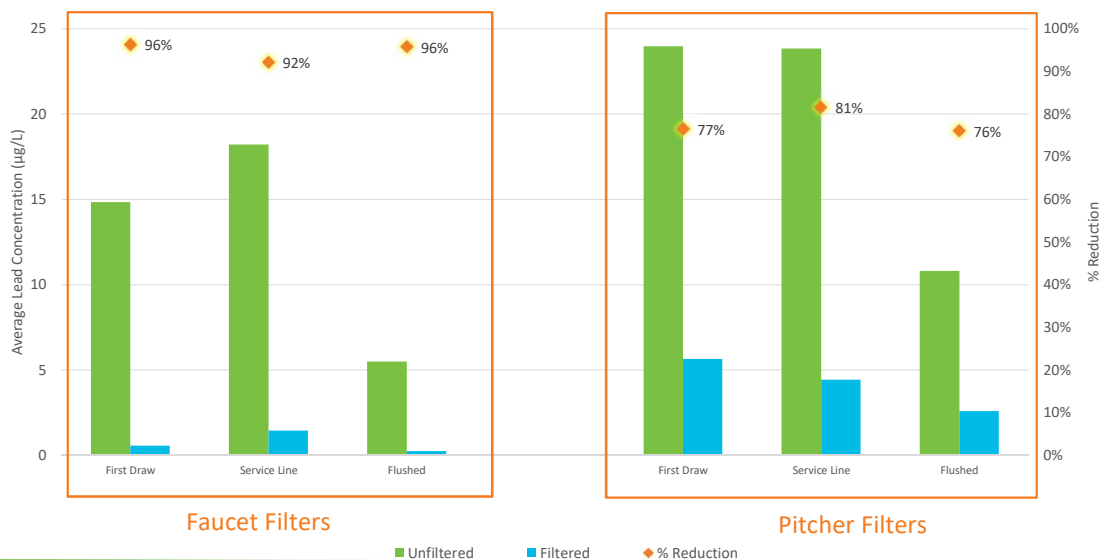
Filter Types – Faucet and Pitcher



38

38

Influence on Filter Type – Faucet vs. Pitcher



39

Combined Influence – Unfiltered Lead Levels and Faucet Filters



Unfiltered Lead Levels	No. Faucet Samples	% Filtered Samples 10 ppb or Below	% Filtered Samples 5 ppb or Below
<= 10 ppb*	379	100%	100%
> 10 and <= 150 ppb*	182	98.9%	92.9%
> 150 ppb	3	0%	0%
Overall	564	99.1%	97.2%

*NSF Certification Test Parameters

40

Combined Influence – Unfiltered Lead Levels and Pitcher Filters



Unfiltered Lead Levels	No. Pitcher Samples	% Filtered Samples 10 ppb or Below	% Filtered Samples 5 ppb or Below
<= 10 ppb*	11	100%	90.9%
> 10 and <= 150 ppb*	16	87.5%	68.8%
> 150 ppb	0	N/A	N/A
Overall	27	92.6%	77.8%

*NSF Certification Test Parameters

41

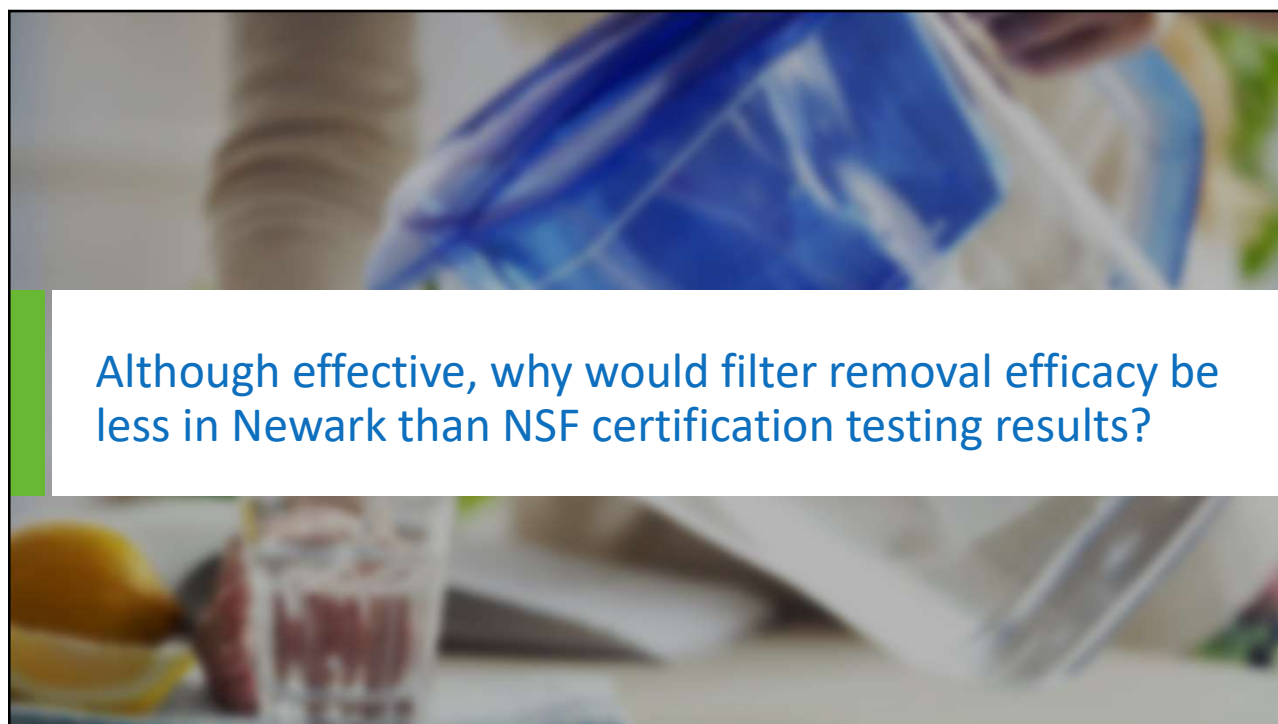
41

Summary – Filter Effectiveness

Samples	No. Filters Properly Installed and Maintained	% All Filtered Samples Below 10 ppb
Stagnated and Flushed Samples	198	97.5%
5 Minute Flushed Samples	198	99.5%

42

42



Although effective, why would filter removal efficacy be less in Newark than NSF certification testing results?

43

Water Quality Differences: NSF Test Water and Newark Water

Parameter	NSF Test Water No. 1	NSF Test Water No. 2	Newark Water
pH	6.5	8.5	7.2 – 7.7
Alkalinity (as CaCO ₃)	10 – 30 mg/L	100 mg/L	29 mg/L
Corrosion Control Treatment	Carbonate chemistry	Carbonate chemistry	Zinc orthophosphate
Lead Levels	150 ppb Assumed all soluble lead	150 ppb Particulate lead = 30% max Fine particulates (0.1 to 1.2 μm) = 20% of particulate	Varies

44

44

Orthophosphate Impact on Filters

- EPA ORD conducted sampling in August 2019
 - Metals, pH, alkalinity, and total inorganic carbon
 - Pb particle size fractionation of service line treated (through POU filter) samples
 - 0.2 μm syringe filters
 - 0.01 μm ultrafilter
 - Nanoparticles ($< 0.01 \mu\text{m}$) found passing through POU filters in 3 of 4 homes
 - Analysis found elemental lead, phosphorus and chlorine (consistent with pyromorphite)



45

45

Key Takeaways

- Filters ***do*** remove lead in Newark
- Filters are ***not certified*** to remove 99% of lead under the NSF 53 certification
- Filters are certified to remove lead to 10 ppb (changed to 5 ppb) under ***specific carbonate water quality conditions***
- Orthophosphate can produce nano-size lead particles that are challenging for the filters to remove—especially pitcher filters
- Flushing prior to filtering reduces lead levels

46

46

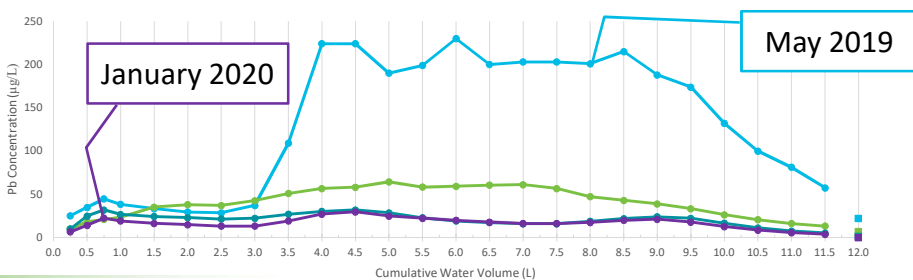
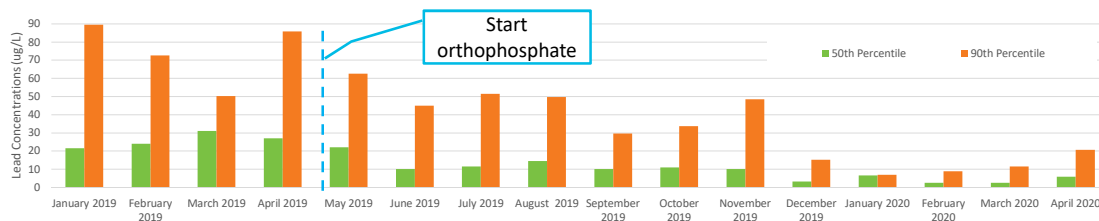
Study Recommendations

- Continue to provide access to filters and cartridges
- Emphasize flushing prior to using filters to reduce lead levels in the unfiltered water
- Use faucet filters where possible
- Public education on flushing and proper install and use of filters
- Continue with parallel efforts to reduce lead levels in the distribution system

47

47

Newark is seeing improvements in CCT



48

48

Newark knows the only permanent solution is to remove the lead



Newark's Lead Service Line Replacement Program Replaces 10,000 Lines in 13 Months

49

49

Thank You!

- City of Newark, NJ
- US Environmental Protection Agency
 - Office of Research & Development
 - Region 2
 - Office of Water
- New Jersey Department of Environmental Protection
- New Jersey Department of Health



50

50

Contact us!

Link to Newark's Filter Study Report:
<https://www.newarkleadsviceline.com/2019filterstudy>



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51

MANAGING MILLIONS OF DATA POINTS

Administering Chicago's Free Lead Testing Program

Sophie Frances Manley, Sanitary Engineer III
City of Chicago Department of Water Management

Kiran Udayakumar, Water Quality Engineer
Arcadis



52

City of Chicago Department of Water Management

- Responsibilities
 - Drinking water treatment
 - Water distribution system maintenance
 - Combined sewer system maintenance
 - Stormwater management
- Serve over 5 million people
 - Residents of the City & 125 suburbs
- Infrastructure
 - 4,311 miles of water main
 - 4,600 miles of sewer main

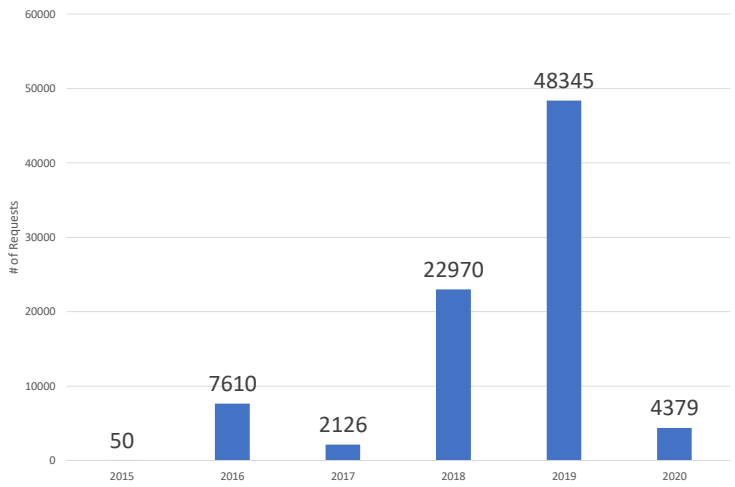


Jardine Water Purification Plant

53

History of Lead Testing

- Free lead testing for Chicago residents since at least 1986
- Change in program in 2016 due to spike in requests
- Customer initiated program through CHI311



Lead Testing Requests per Year since 2015

54

Current Program

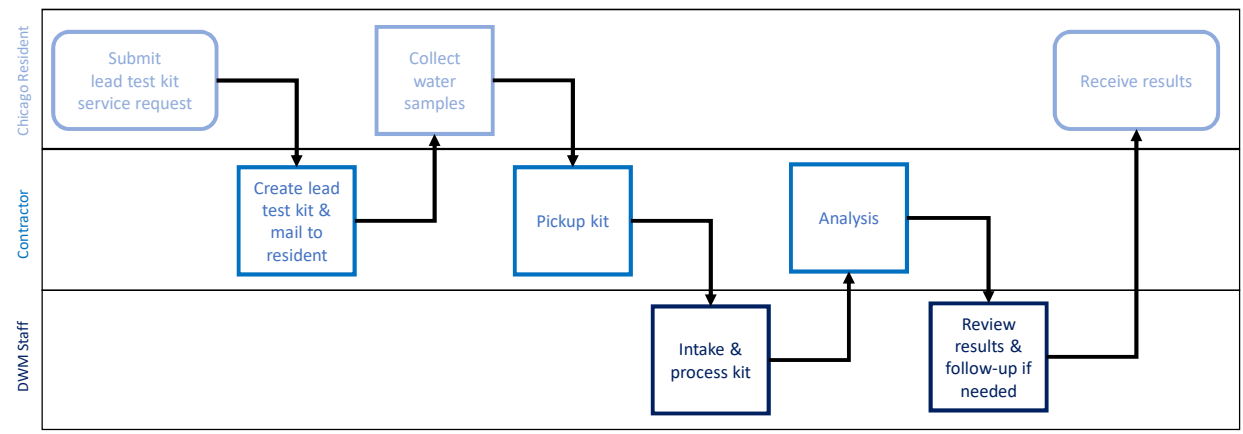


Lead Test Kit

- Voluntary program for Chicago residents
- Two Options:
 1. **Lead Test Kit**
 - A test kit is mailed to the resident
 - ~89% of requests
 2. **Lead Test Visit**
 - A DWM representative goes to the resident's home to complete the kit for them
 - ~11% of requests

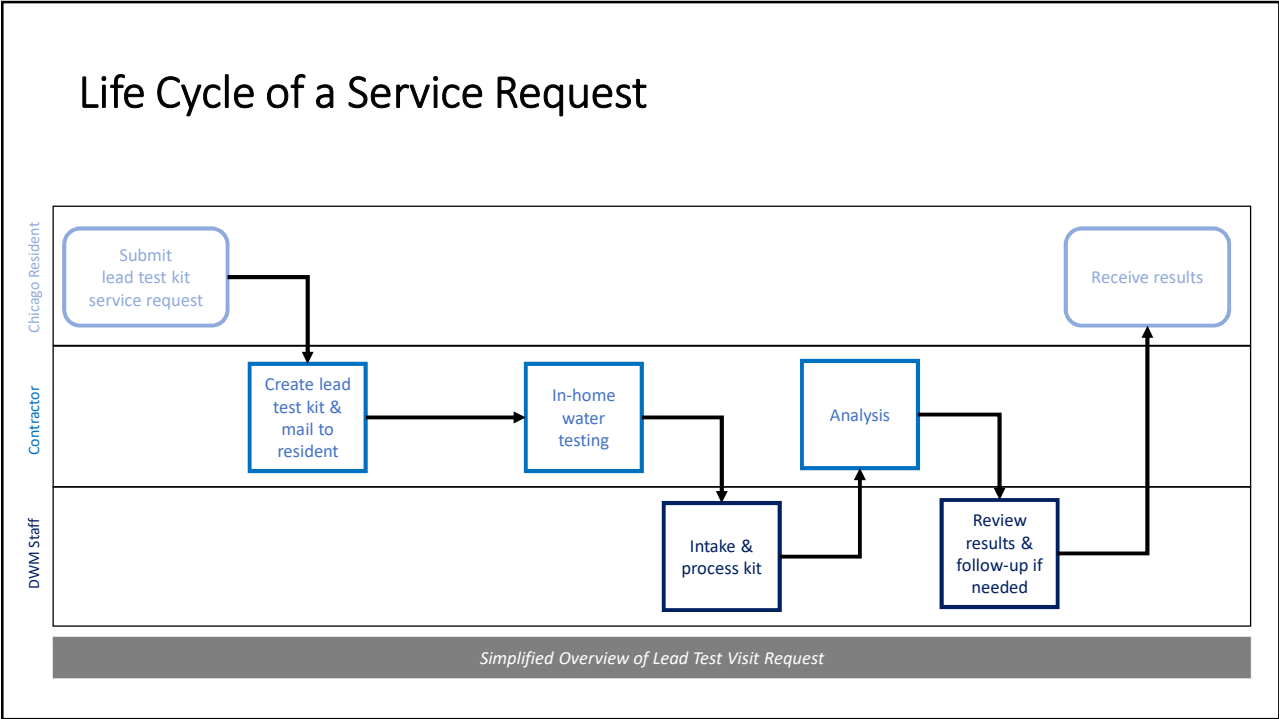
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Life Cycle of a Service Request

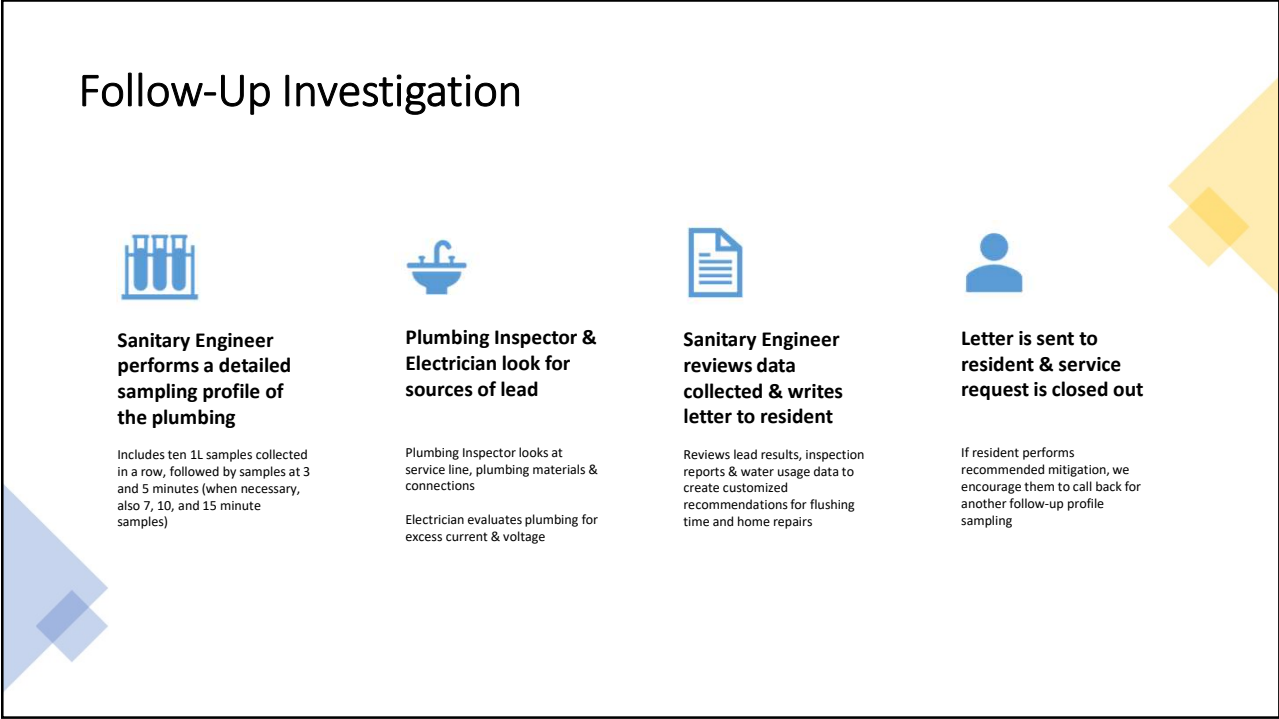


Simplified Overview of Lead Test Kit Request

56



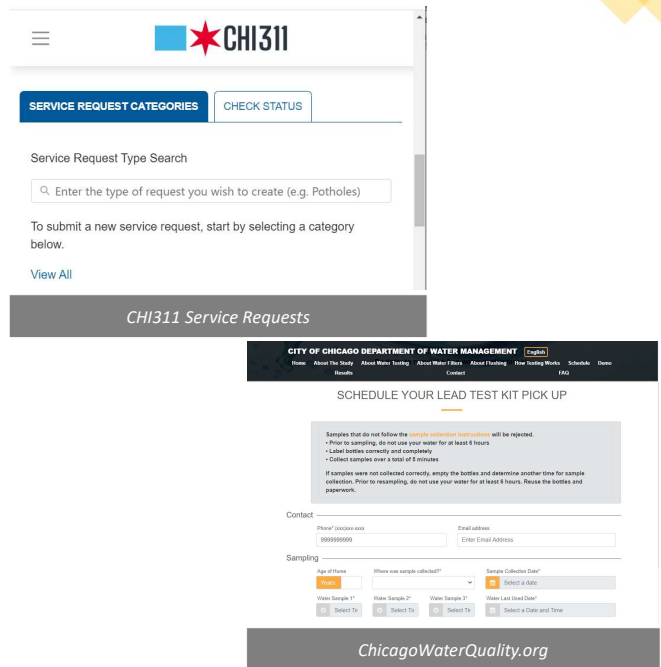
57



58

Data Collection

- 1. **CHI311** non-emergency services ticketing system
 - Based on Salesforce Customer Relationship Platform (CRM)
- 2. Integrated to localized data management solution **InforEAM**
- 3. Customized **Water Quality Website** for resident data collection and scheduling
- 4. Customized **Geographic Information System (GIS) Applications** for field data collection
- 5. Result data from **Laboratory** after analysis




59

How many data points are collected?

Millions of data points!


60

Reporting Needs



Program Stakeholders

- Chicago resident
- Engineers and researchers
- Utility management
- Elected officials
- Regulators



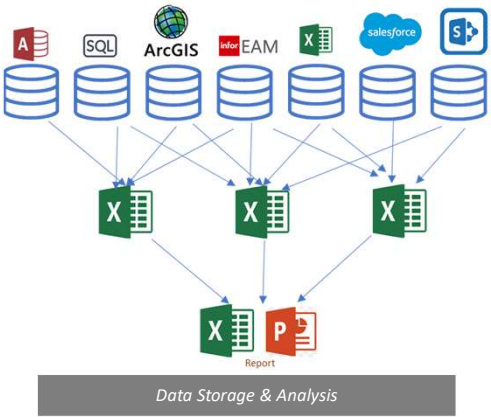
What do they want to know?

- Lead test results
- Program fulfillment
- Results over time
- Trends
- Proof of contact

61

Data Storage & Analysis

- Current data point numbers and storage locations
 - Over 85,000 individual requests
 - Over 90 data fields
 - Over 7 different data storage locations
- Variety of unique analysis methods and reports
- Drilling through data and identifying trends with large data
- Data interpretation and analysis was tedious and time consuming with increasing requests.
- Recent implementation: Dashboards

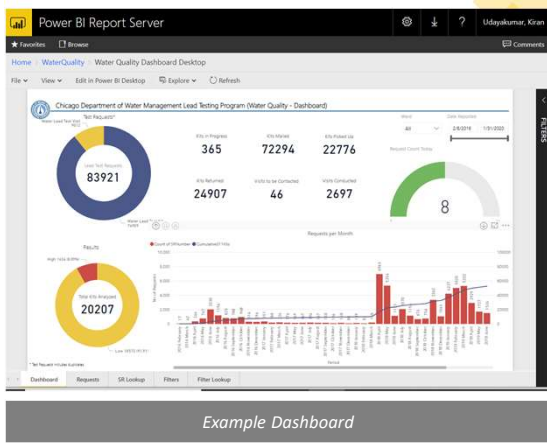


Data Storage & Analysis

62

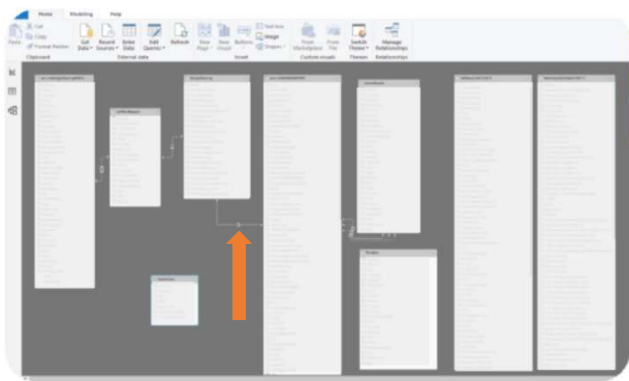
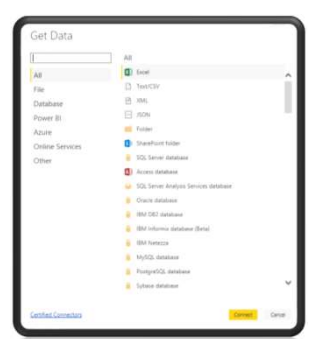
Business Intelligence Dashboard Solution

- Large datasets
- Access multiple data sources
- Real time data updates and analysis
- Drill through data
- Mobile access
- Notifications



63

Large Datasets & Multiple Data Sources



...and many more!

64

Drill through Data

Filter selection panels

Filter selection through other visuals

Extract filtered data into reports

65

Dashboard Sharing and Real-Time Updates

- Embedded on collaborative platforms
- User access privileges
- Data drilling and exporting access
- Schedule refresh rate
- Updates mobile and desktop
- Unique Mobile application
- Mobile Notification

66

Dashboard Sharing and Real-Time Updates

- Real time data analysis and notification to required personnel
- Streamlined Data Distribution
- Dashboard showed a 360 view of the program
- Improved workflows and productivity
- Identified regions of concern
- Easily access multiple data sources at same platform



67

Implementation and Security

- On premise Implementation
- Multi-tier enterprise architecture
- Private network
- Access to legacy data sources and flat files
- Single Sign on
- Active Directory Authentication



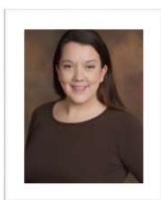
68

Key Takeaways

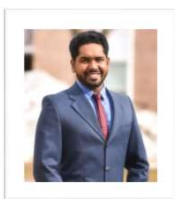
1. When creating or expanding lead testing programs, keep in mind future reporting needs to help determine what data needs to be collected
2. Data management can get tedious over the course of the program
3. Digital dashboards can be a helpful solution and aid in making real-time, data-driven decisions

69

Thank you!



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70

Assessment of Lead in Childcare and School Drinking Water

John E. Tobiason, PhD, PE, BCEE

(tobiason@umass.edu)

*Professor, Dept. of Civil & Environmental Engineering
University of Massachusetts at Amherst*

**AWWA Webinar
Managing Risk from Lead Service Lines
26 May 2020**

Tobiason, Lead in School Drinking Water, AWWA Webinar May 2020

71

71

Science: Lead (Pb) Health Concerns

- **Long known that:**
 - **Pb is a neurotoxin, accumulates in tissues/bones, teeth.**
 - **Pb can cause cardiovascular diseases, brain damage, carcinogenic properties, lead poisoning.**
 - **Vulnerable populations are pregnant women, developing fetuses, infants and children.**
- **The US CDC lowered the “level of concern” for children’s blood lead level (BLL) from 10 µg/dL to 5 µg/dL in 2012. Medical treatment is recommended at levels > 45 µg/dL**
- ***Frequent statements by public health officials that there is no safe level of lead exposure for children***

Tobiason, Lead in School Drinking Water, AWWA Webinar May 2020

72

72

Policy: Pb Sources & Human Exposure

- **Advantages of using lead in materials/products took precedence over health/exposure concerns**
 - **Why? Insufficient knowledge/data? Economics?**
- **Human exposure to lead: mostly due to lead paint, contaminated soil/dust. US society took action!**
 - **Lead paint was banned in 1978.**
 - **Leaded gasoline phased out in mid 1970's**
- **Pb exposure from consumed drinking water**
 - **Decisions to use Pb containing materials to convey water from the water main to consumers**
 - **Relative contribution increasing as other sources decrease**

Tobiason, Lead in School Drinking Water, AWWA Webinar May 2020

73

73

Plumbing: Pb Sources for DW

- **Plumbing components**
 - **Old small diameter water mains**
 - **Lead service lines (LSLs), lead goosenecks, copper piping**
 - **Building plumbing: brass fixtures, solder (before 1986) containing lead, copper & galvanized iron piping**



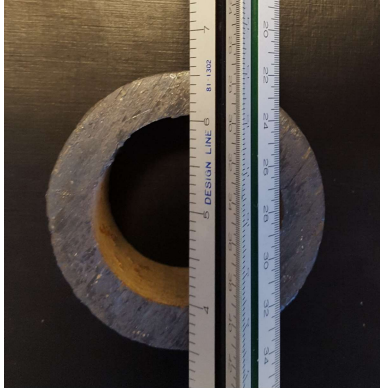
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74

74



Typical ~ 3/4 inch lead service line removed from ground, coiled (material is flexible!)



Piece of large ~ 2 inch inside diameter lead service line (courtesy Steve Price, Denver Water)

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75

75



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76

76

Policy: Regulating Pb in Plumbing Materials-USA

- **1986 SDWA prohibited use of pipes, solder or flux that were not “lead free”; at that time, “lead free” defined as < 0.2 % for solder and flux, and < 8% for pipes (by weight)**
- **1996 SDWA required plumbing fittings and fixtures to be in compliance with voluntary lead leaching standards**
- **2011 Reduction of Lead in Drinking Water Act (RLDWA) re-defined “lead free” to be weighted average across wetted surface of < 0.25% lead by weight, but eliminated compliance with voluntary lead leaching standard**
 - **Prohibited introduction of products that are not lead free**
 - **Exemptions for variety of products not usually used to provide drinking water**
 - **NSF/ANSI 372 standard as of Jan 2014**

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77

77

Science: Pb from materials into water

- **Zero valent solid Pb (and Cu) metals in plumbing materials get oxidized (corroded)**
 - Corrosion products include dissolved and particulate (solid) forms of the metals; these can enter drinking water
- **How much and how fast? Complex!**
 - Redox reactions, equilibrium reactions, solid phases, electrochemistry, microbiology, etc.....
- **Extensive research on various critical factors:**
 - pH, inorganic carbon, phosphate, other inorganic anions, oxidants, temperature.....
 - dissimilar metal connections...
 - hydraulics: velocity, stagnation....
- **Chemistry & operation can limit, but not eliminate, Pb transfer to water from Pb containing materials**

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78

Policy: What level of Pb in Water is OK?

- **US EPA Maximum Contaminant Level Goal (MCLG) of zero for drinking water (due to cancer endpoint)**
- **Prior to 1991 US EPA Lead and Copper Rule (LCR), lead (Pb) regulated at 50 ppb at entry to distribution system**
- **1991 US EPA LCR set an “Action Level” (AL) of 0.015 mg/L**
 - If more than 10% of tap water samples from homes > AL, public water supplier must take “action” (education, service line removal, optimal corrosion control).
- **World Health Organization: guideline value of 0.010 mg/L**
- **Health Canada maximum acceptable concentration (MAC) for total lead at the tap:**
 - March 2019, set at 0.005 mg/L (was 0.010 mg/L)
- **American Academy of Pediatrics (2016)**
 - Lead not greater than 1 ppb (0.001 mg/L)

79

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79


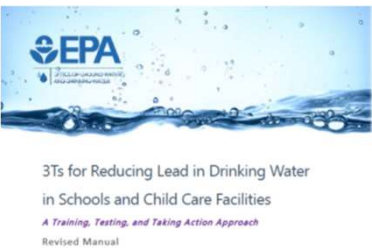
Policy: Pb/Cu in School tap water

- **US EPA 1988 Lead Contamination Control Act (LCCA)**
 - US EPA established a voluntary program aimed at decreasing the lead (and copper) concentrations in school drinking water
 - Applies to K-12 schools and early education and care (EEC) programs
 - Provided a list of banned water coolers due to lead materials
 - Provided guidance on how and where to collect water samples
- **US EPA “3Ts” guidance manual provides many details on fixture sampling and follow-up actions.**
 - Original 3Ts: training, testing, telling
 - **New revised 3Ts Guidance Oct 2018**
 - October 2018: training, testing, taking action

80

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Oct 2018 Revised 3Ts:

- “no safe level of lead for children”
- “..reduce their lead levels to the lowest possible concentrations.”
- No longer refer to an “action level” for lead (was 20 ppb) for school drinking water
- Indicates that there may be a state or local “remediation trigger level” to refer to
- Thus, states must decide and give guidance
- May 2019: MassDEP issued guidance to schools/EECFs for Pb < 1 ppb

https://www.epa.gov/sites/production/files/2018-09/documents/final_revised_3ts_manual_508.pdf

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81

81

MassDEP/UMass Assistance Program

- **Late April 2016: Governor Baker administration announces the “Massachusetts Assistance Program for Lead in School Drinking Water” to fund implementation of LCCA based sampling of taps at K-12 public schools and EECs in Massachusetts**
 - \$2.75 M from the Massachusetts Clean Water Trust
 - \$0.97 M from USEPA WIIN Act Funding to states
 - Implemented by MassDEP and UMass Amherst
 - Extensive involvement of MWRA
 - Supported by MassDPH, MassDESE, MassDEEC, PWSs
 - Phase 1: 5/2016 – 3/2017; Phase 2: 8/2017-06/2020
 - Phase 3: Extended Assistance Program 1/20 – 6/21
- **DEP funds UMass Amherst to implement Program**
 - Technically competent, 3rd party, PIs to manage project
 - Much easier for UMass to hire hourly, non-benefited, staff
- **UMass Project Managers and Technical Assistance Providers worked closely with DEP staff to develop & implement program**

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82

82

Program Components

- Existing DEP LCCA Program
- Forms and information materials (on DEP Website)
- Application by school system
- Informational Meeting w/ Community
- Sample Plan/Fixture Map
- Web-Based Lead and Copper Management Tool
- Sampling
- Laboratory Analyses
- Reporting of Lab Results to DEP and Schools
- Follow-up Steps

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83

83

Summary of MassDEP Phase 1&2 Program (2016-2018)

- Sampled ~ 985 school building in 190 municipalities
- Range of 1 to 76 buildings per school system
- Average of 39 sample locations (1 – 234 range) per building
- Average of 69 samples per building
 - 250 mL primary (first draw) after 8-18 hrs stagnation, all locations
 - 250 mL flush sample after 30 seconds flushing (most locations)
- ~70,000 samples collected (~63,000 analyzed by commercial labs), cost of \$1.8 M (lead and copper for all samples)
- **Final report on Phase 1 Assistance Program issued 2 May 2017**
<http://www.mass.gov/eea/agencies/massdep/water/drinking/testing-assistance-for-lead-in-school-drinking-water.html>
- **Phase 2 & overall reports: 2020**

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84

MassDEP EAP (2020-2021) & SWIG

- **MassDEP awarded \$0.96 M USEPA WIIN Act Funding**
 - Focus on children < 6 years of age
 - Sampling and analysis for lead only (not copper)
 - Accepted first round of applications early 2020.
 - 65 facilities accepted into program; 58 public & private group childcare facilities, 7 schools
 - Sampling & analysis on hold due to COVID-19
- **SWIG: New MA Clean Water Trust funded program**
 - **School Water Improvement Grants**
 - \$3000/fixture grants for purchasing and installing point-of-use filtered water bottle filling stations
 - Replacement of fixtures that showed > 1 ppb lead in testing conducted in accordance with MassDEP/UMass Assistance Program (LCCA Program)
 - Accepted applications online early 2020, Round 1 awards announce 4/29/2020. 32 school districts, 225 fixtures, \$675,000
 - UMass technical assistance on assessing applicant eligibility, appropriate fixtures to replace (hallway bubblers (fountains) mostly).

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85

85

What to Sample (From DEP UMass Sampling Training Document)



P-P-F

Paired drinking water bubblers and some classroom sinks share a main water pipe that splits to provide water to two or three fixtures.

A primary (P) sample is taken from all fixtures (each fixture has its own location code) but only one flush (F) sample is taken.



P-P-F

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86

86

What to Sample (continued)



**Kitchen Kettle
(Cold only)**

Ice Machine

**Food
Preparation Sink**

These have one main water line. The sequence for sampling is P-F

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87

87

What to Sample (continued)



**Hallway Water Cooler
(sometimes two water
coolers side-by-side,
each gets Primary and
Flush samples)**

**Nurse's Office
Sink (cold only)**

**Teacher's Lounge Sink
(cold only)**

all are P-F

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88

88

Filtered Fountain & Bottle Filling Station



MA SWIG Program providing grants (\$3000/fixture, installed) for schools to replace fountains with Pb > 1 ppb based on appropriate testing procedures

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89

89

Laboratory Analysis Result Reporting

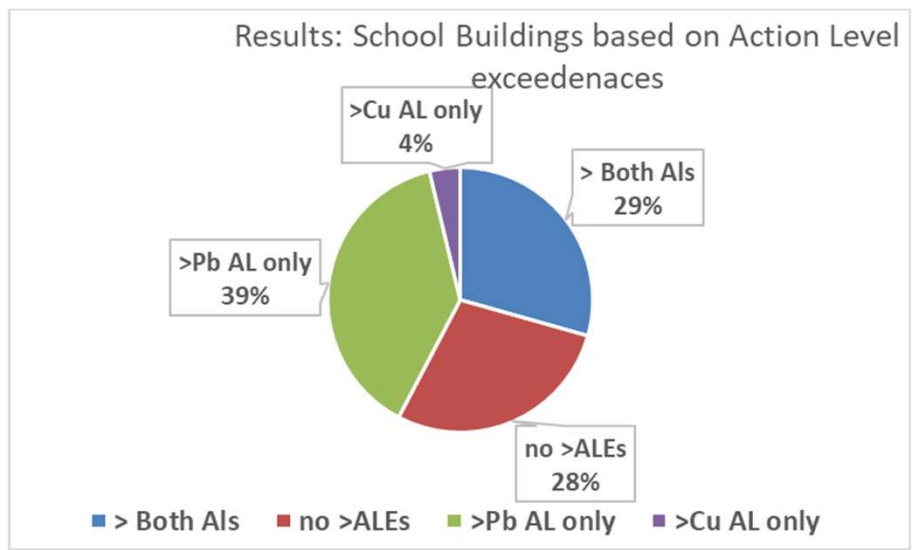
- **Laboratories reported all analytical results electronically to MassDEP (eDEP system)**
 - Performed only by Massachusetts DEP-certified laboratories that were e-DEP compliant
- **MassDEP Phase (1) or UMass (Phase 2) emailed the analytical results (attached Excel file) to school system (1 to several schools at a time) along with DEP contacts, information links, and template letters for parents**
- **DEP transferred the Sampling Results to the online LCCA Management Tool**
- **DEP posted results on public website ~ 2 weeks after sending to schools (see website below for all MA LCCA data) (<https://eeasonline.eea.state.ma.us/Portal/#!/search/leadandcopper>)**
- **MassDPH followed-up with an email with information about Pb and Cu and health and additional guidance**

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90

90

Overall Results: School Buildings Basis (974 schools)

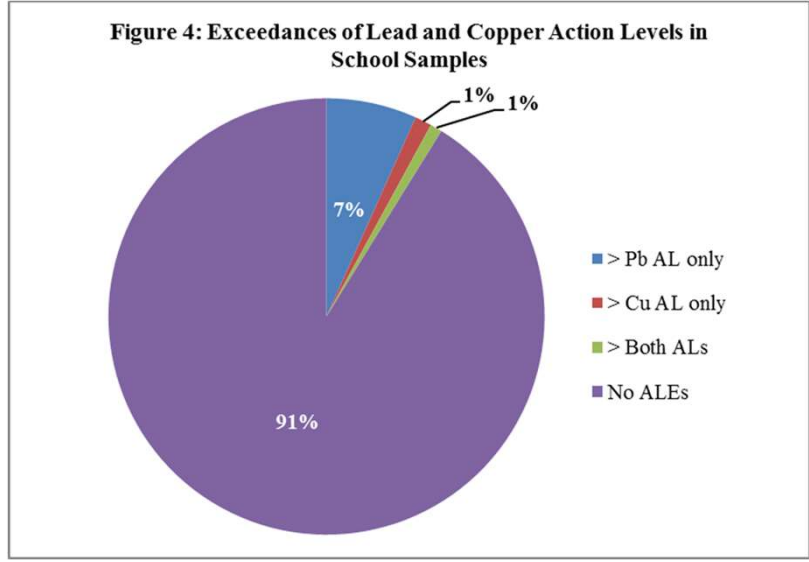


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91

91

Action Level Exceedences: Number of samples basis



Source: MassDEP Assistance Program Final Report, May 2017

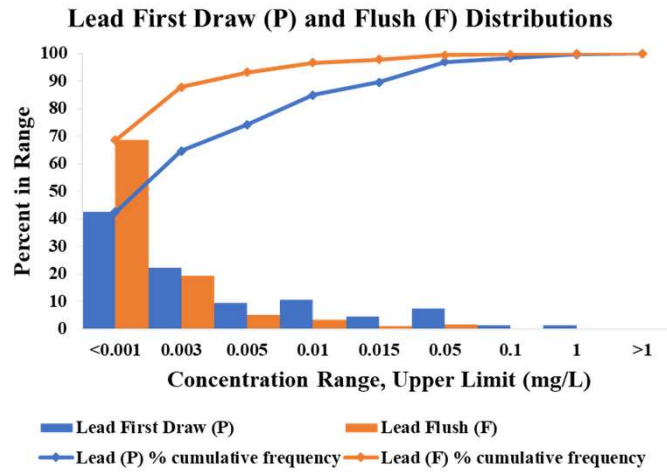
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92

92

Results, Sample Basis: Lead Concentration Distribution

First Draw Samples: ~ 11 % > 15 ppb, 26 % > 5 ppb, 58% > 1 ppb
Flush Samples: ~ 2.1 % > 15 ppb, 6.9% > 5 ppb, 31% > 1 ppb
- Significant impact of 30 sec flush on decreasing Pb levels

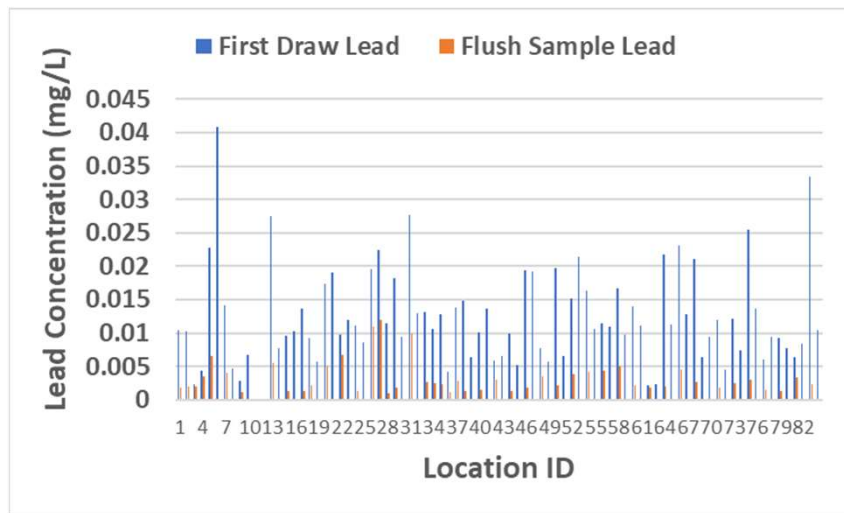


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93

93

One school: large impact of 30 sec flush on lead levels



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94

94

Remedial Actions by Schools

- MassDEP has these recommendations:
- **Contact Local Public Water System and MassDEP Drinking Water Program for assistance**
- **Immediate Measures**
 - Shut Off Problem Fixtures
 - Implement a Flushing Program (track via Manual Flushing Log) (this is a temporary measure, helpful, not a solution)
- **Conduct Outreach to Staff and Parents**
 - Transparency is critical
- **Determine if the source of the contamination is the fixture or the plumbing**
 - Check Plumbing Profile
 - Possibly replace plumbing
 - Install POU lead removal treatment (focus of SWIG program)
 - Follow-up Sampling & Analyses
- **Develop Plan of Permanent Measures**
- **Report remedial actions taken on the MassDEP online Management Tool**

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95

95



Old school fixtures

New ("Pb free") school fixtures
(NSF/ANSI 372 since 2014)



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96

96

Control of Pb & Cu Levels Premise Plumbing

- **Removal/elimination of all Pb and Cu from water system materials:**
 - Very challenging due to cost, but significant progress has been made, and more needs to be done (LSLs, premise plumbing)
- **Source water treatment to remove Pb & Cu? NOT the issue, very rarely the source.**
- **Source water treatment to minimize corrosion of materials containing Pb and Cu – YES, very important**
 - pH, alkalinity (DIC), phosphate, oxidants, chloride/sulfate, etc.
 - Use optimal corrosion control treatment (OCCT)
- **Flushing of water fixtures by consumers prior to consumption, drawing cold water for consumption – YES, but requires consumer education; unknown duration of effective impact**
- **Point of use (POU) treatment for Pb removal: significant implementation (e.g., Flint, MI; schools; other), O&M ?**
- **US EPA LCR and guidance, and public health guidance, reflect all these measures**

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97

97

MassDEP Assistance Program Findings

- It is common (69%) for a school building to have at least one sample that has lead > 15 ppb; very rare for a school to have all samples < 1 ppb lead
- Most of the sample AL exceedance results for Pb are for the primary or first draw sample (11%) versus flush sample (2.3%), highlighting the short-term benefits of flushing.
- Exceedances of the copper Al of 1.3 mg/L are infrequent and similar for first draw (3.2%) and flush (1.8%) samples. Elevated copper levels are more systematic, and could possibly be controlled by optimum corrosion control.
- *Important to sample all fixtures in a building as not possible to predict which fixtures are a problem, i.e., there are not usually “representative” fixtures for a building (unless all are below detection limits!).*

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98

98

Acknowledgements (1): who did the work!

- **UMass Amherst based Project Team**
 - John Tobiason, David Reckhow, *co-Principal Investigators*
 - Rick Larson, Bob Hoyt, Marie-Francoise Hatte, *co-Program Managers*
 - Tom Bienkiewicz (MRWA), Gene Brunelle, Harrison Corbett, Jim Dillon, Kate Gallagher, Mark Griffin, Marie Iken, Ted Kenney, Lynda Laine (MRWA), Thom Martens, Heather Minott, Wayne Southworth, Anita Wolovick, *Technical Assistance Providers*
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- **Commonwealth & MassDEP Team (a few of the many!)**
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 - Yvette Depeiza, Damon Gutterman, Andrew Durham, Margaret Finn, Ken Pelletier, Tio Yano, Nicole Della Porta, Jessica Sibirski
- **All the school, PWS, and other municipal employees**

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99

99

Acknowledgements (2): who did the work!

- **Massachusetts Water Resources Authority**
 - Sample bottles, sample analyses (~8000 for this Program)
 - MWRA has conducted many more analyses for member communities outside of this program, at no charge to those communities
- **Commercial Laboratories**
 - 12 different laboratories
 - provided important communication, sample bottles, analysis
 - Large workload, short time period

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100

100

Questions?

Thank you for your attention!

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101

101

Ask the Experts



Sandra Kutzing
CDM Smith



Carol Rego
CDM Smith



Sophie Manley
City of Chicago
Department of
Water Management



Kiran Udayakumar
Arcadis



John Tobiason
University of
Massachusetts at
Amherst

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Please specify to whom you are addressing the question.

102



102

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103



103

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- As part of your registration, you are entitled to an additional 30-day archive access of today's program.

- Until next time, keep the water safe and secure.

104



104

Presenter Biography Information

- Sandra Kutzing is a Principal Engineer at CDM Smith with 18 years of experience in drinking water. Ms. Kutzing has a BS from the University of Illinois in Civil Engineering and an MS from the University of Washington in Civil Engineering. For the past few years, Ms. Kutzing has been working with utilities specifically on optimizing corrosion control and establishing and managing lead service line replacement programs.
- Carol Rego is among the water industry's leading experts in water quality and corrosion control treatment. With almost four decades of experience as a water supply and treatment specialist at CDM Smith, Carol's expertise spans treatability studies, process selection and drinking water research, water supply planning, water treatment plant evaluation, operations, and design, as well as Safe Drinking Water Act (SDWA) and Lead and Copper Rule (LCR) compliance.
- Sophie Frances Manley: Sophie graduated from the University of Notre Dame with a degree in Environmental Engineering. Since 2016, she has been working with the City of Chicago Department of Water Management as an engineer in the Division of Water Quality. She is passionate about working for a water utility.
- Kiran Udayakumar: Kiran graduated with a Master's Degree in Environmental Engineering from Michigan Technological University. He works for Arcadis as a Water Quality Engineer in the Water Business Line. He is experienced in water quality, treatment & corrosion control and incorporating digital solutions in his work.
- John E. Tobiasson is Professor and Department Head of Civil and Environmental Engineering at the University of Massachusetts at Amherst. BS in Civil Engineering University of New Hampshire (1976), MS in Environmental Engineering University of North Carolina at Chapel Hill (1979), PhD in Environmental Engineering Johns Hopkins University (1987). Past President of AEEESP and past Chair of the AWWA Water Science Research Division Board of Trustees. He is a registered Professional Engineer (NH), and a Board Certified Environmental Engineer by the AAEES.

105



105

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106

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107



107

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